

Stereoselective Synthesis of 3-Hydroxyazetidines via Regioselective Halogenation of 2,3-Epoxyamines by Using Magnesium Bromide

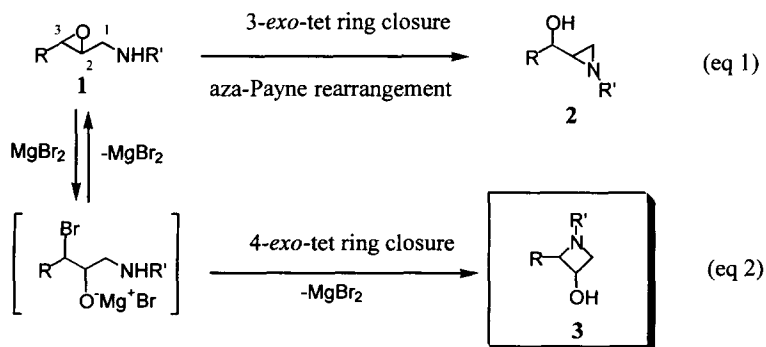
Michinori Karikomi*, Kouji Arai and Takashi Toda

Department of Applied Chemistry, Faculty of Engineering,
 Utsunomiya University
 Ishiicho 2753, Utsunomiya 321, Japan

Abstract: A novel stereoselective synthesis of 3-hydroxyazetidines starting from 2,3-epoxyamines is described. Regioselectivity of the intramolecular cyclization of 2,3-epoxyamines is controlled by the use of magnesium bromide. The cyclization proceeds with retention of the stereochemistry, caused by double inversion of two sequential S_N2 reactions of the epoxyamines. © 1997 Elsevier Science Ltd.

Azetid-3-one derivatives are interesting as the non-hydrolyzable isosteres of β -lactams.¹ Although 3-hydroxyazetidines seem to be good precursors for the azetid-3-ones,² there have been only few examples for efficient synthesis of these compounds.³

On the other hand, intramolecular cyclization of 2,3-epoxyamines (1) leading to hydroxymethylaziridines (2) is known as the aza-Payne rearrangement (eq 1). This reaction proceeds through the 3-*exo*-tet type cyclization.⁴ If the nitrogen reacts at the terminal C-3 carbon, 3-hydroxyazetidines (3) will be formed. However, the 4-*endo*-tet-like type cyclization reaction have not yet been reported, even though the case of the terminal C-3 carbon is activated by conjugate effects.^{4a} Clearly, a method which allows for the control of both regio- and stereochemistry would be of great utility for the synthesis of these compounds. Herein, we report that magnesium bromides can be used effectively for the conversion of *cis*- and *trans*-2,3-epoxyamines to *trans*- and *cis*-3-hydroxyazetidines, respectively (eq 2).



Scheme 1

2,3-Epoxyamines (**1**) were prepared from 2,3-epoxyhalides according to the previously established method^{5a} and to the modified known methods.^{5a,c}

Preliminary investigations were carried out in dioxane using magnesium bromide etherate as the brominating reagent and *N*-benzyl-*N*-(*cis*-3-phenyl-2,3-epoxypropyl)amine (**1a**) as the substrate. The reaction has to be performed in the presence of 5 mol% of a magnesium bromide etherate (Table 1, Entry 2) with good regio- and stereoselectivity. The stereochemistry of the cyclization proceeded with retention,⁶ because halogenation of the epoxide with magnesium bromide proceeded with inversion and the following displacement of the halogen by the amino group also proceeded with inversion. In a further experiment, we tested the efficiency of the halogenating reagent. Use of magnesium bromide etherate and magnesium bromide gave similar results (Entry 2 and 4). In the case of addition of 1 mol% of magnesium bromide etherate, it was seen that the reaction was retarded and the ratios of the hydroxymethylaziridines (**2a** and **2b**) were increased (Entry 3). Magnesium iodide is also effective, and the reaction was completed in 1h (Entry 5).

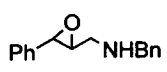
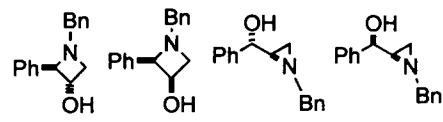
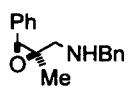
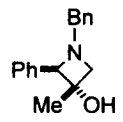
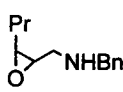
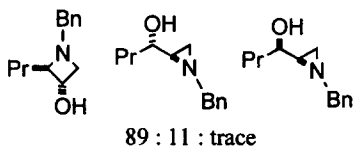
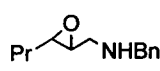
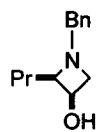
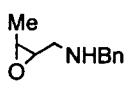
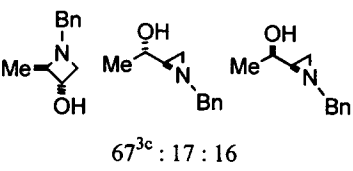
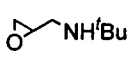
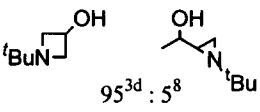
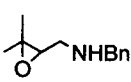
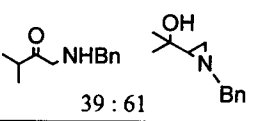
Table 1. Synthesis of *trans*-3-Hydroxy-2-phenylazetidine

Entry	MgX ₂ (mol%)	Time/h	Yield/% ^{a)}	Product /Ratio (3a : 3b : 2a : 2b) ^{b)}
1	MgBr ₂ -Et ₂ O (50)	5	48	88 : 4.1 : 7.1 : 0.5
2	MgBr ₂ -Et ₂ O (5)	5	69	89 : 1.9 : 8.9 : trace
3	MgBr ₂ -Et ₂ O (1)	48	60	61 : trace : 32 : 7.5
4	MgBr ₂ (5)	5	66	86 : 9.7 : 4.0 : none
5	MgI ₂ (5)	1	59	85 : 9.0 : 6.5 : none

^{a)} Isolated yield. ^{b)} Ratio was determined by ¹H NMR.

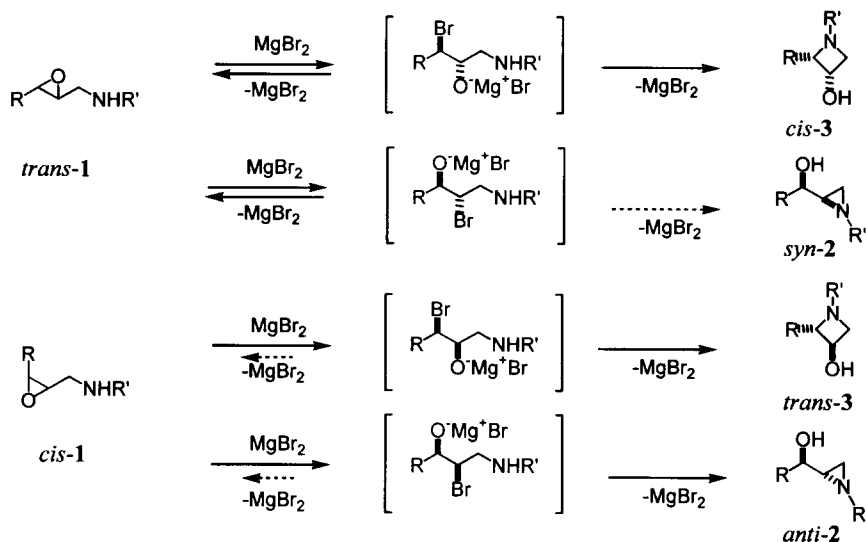
Following these observations, the procedure was extended to various 2,3-epoxyamines (**1**). As shown by the examples compiled in Table 2, reasonable yields of the expected 3-hydroxyazetidines (**3**) were obtained (Table 2, Entry 1-8), except in the case of a dimethyl substituted derivative, which gave hydroxymethylaziridine (**2**) as the main product and a ketone derivative as a minor product (Entry 9). Also, when we applied magnesium iodide as the halogenation reagent for *N*-benzyl-*N*-(*trans*-3-phenyl-2,3-epoxypropyl)amine, *trans*-3-hydroxyazetidine (**3a**) was obtained as an unexpected stereoisomer (Entry 3). Thus, magnesium iodide is not effective for the stereospecific synthesis of **3**. In the case of *cis*-2,3-epoxyamines, the formation of small amounts of the corresponding hydroxymethylaziridines (**2**) as side-products was observed (Entry 5 and 7). Magnesium bromide is well known as an effective halogenating reagent for epoxides.⁷ Accordingly, the rapid formation of β -halohydrine intermediates should be expected by the attack of magnesium halide, and the next cyclization step would be the rate-determining step. In the case of *trans*-2,3-epoxyamines (**1**), the rapid equilibration between the starting materials and the β -halohydrine intermediates will take place. On the other hand, equilibration of β -halohydrines and *cis*-2,3-epoxyamine (**1**) will not occur or is very slow, because the *cis*-substituted groups are not preferred for the reproduction of the epoxide ring. Thus, a small amount of hydroxymethylaziridines is formed (Scheme 2).

Table 2. Synthesis of 3-Hydroxyazetidines^{a)}

Entry	Substrate	MgX ₂ (5 mol %)	Time h	Yield ^{b)} %	Product Ratio ^{c)}
					
1		MgBr ₂ -Et ₂ O	8	79	6.1 ^{3b} : 94 : trace : trace
2		MgBr ₂	3	73	6.5 ^{3b} : 92 : trace : 1.3
3		MgI ₂	2	52	93 ^{3b} : 7.0 : none : none
4		MgI ₂	24	77 ^{3b}	
5		MgBr ₂ -Et ₂ O	12	78	
6		MgBr ₂ -Et ₂ O	9	58	
7		MgBr ₂ -Et ₂ O	8	31	
8		MgBr ₂ -Et ₂ O	3	69	
9		MgBr ₂ -Et ₂ O	10	58	

^{a)} All reactions were conducted in refluxed anhydrous 1,4-dioxane. ^{b)} Isolated Yield.

^{c)} Ratios were determined by ¹H NMR spectroscopy.



Scheme 2

A typical procedure for the synthesis of *trans-3a* is as follows. *N*-Benzyl-*N*-(*cis*-3-phenyl-2,3-epoxypropyl)amine **1a** (239 mg, 1.0 mmol) and magnesium bromide etherate (12.8 mg, 0.05 mmol) are refluxed in dry dioxane (10 ml) for the consumption of **1a**, monitored by TLC. The reaction mixture was passed through silica-gel short column chromatography with ethylacetate as an eluent. The eluted mixture was distilled using Kugelrohr apparatus. The products were almost pure and were found to consist of *trans-3a*, *cis-3b*, *anti-2a*, and *syn-2b* in a 89 : 1.9 : 8.9 : trace ratio by ^1H NMR analyses.

In summary, a broadly applicable and simple procedure for the efficient cyclization of 2,3-epoxyamines (1) to 3-hydroxyazetidines (3) has been developed. Research is ongoing to investigate the application of this methodology to other heterocyclic systems.

REFERENCES

- Podlech, J.; Seebach, D. *Helv. Chim. Acta*, **1995**, *78*, 1238-1246.
- Kozikowski, A. P.; Fauq, A. H. *SYNLETT*, **1991**, 783-784.
- For a review, see: (a) Cromwell, N. H.; Phillips, B. *Chem. Rev.*, **1979**, *79*, 331-358. (b) Toda, T.; Karikomi, M.; Ohshima, M.; Yoshida, M. *Heterocycles*, **1992**, *33*, 511-514. (c) Higgins, R. H.; Faircloth, W. J.; Baughman, R. G.; Eaton, Q. L.; *J. Org. Chem.*, **1994**, *59*, 2172-2178. (d) Gaertner, V. R. *Tetrahedron Lett.*, **1992**, *39*, 4691-4694.
- (a) Najime, R.; Pilard, S.; Vaultier, M. *Tetrahedron Lett.*, **1992**, *33*, 5351-5354. (b) Moulines, J.; Charpentier, P.; Bats, J-P.; Nuhrich, A.; Lamidey, A-M. *Tetrahedron Lett.*, **1992**, *33*, 487-490. (c) Ibuka, T.; Nakai, K.; Habashita, H.; Hotta, Y.; Otaka, A.; Tamamura, H.; Fujii, N.; Mimura, N.; Miwa, Y.; Taga, T.; Chounan, Y.; Yamamoto, Y. *J. Org. Chem.*, **1995**, *60*, 2044-2058.
- (a) Karikomi, M.; Yamazaki, T.; Toda, T. *Chem. Lett.*, **1993**, 1787-1790. (b) Gaertner, V. R. *Tetrahedron*, **1967**, *23*, 2123-2136. (c) Asensio, G.; Mello, R.; B-Bernardini, C.; Elena, M.; Núñez, G.; Castellano, G. *J. Org. Chem.*, **1995**, *60*, 3692-3699.
- 3-Hydroxyazetidines (3) and hydroxymethylaziridines (2) were also synthesized by the modified procedure according to Ref. 3b, 3c, and 4a.
- For a review, see: (a) Bonini, C.; Righi, G. *Synthesis*, **1994**, 225-238. Regiocontrolled cyclization of epoxy compounds has been reported by use of magnesium halide. See: (b) Babler, J. H.; Bauta, W. E. *Tetrahedron Lett.*, **1984**, *25*, 4323-4324. (c) Cooke Jr., M. P.; Houpis, I. N. *Tetrahedron Lett.*, **1985**, *26*, 3643-3646. (d) Wu, T-C.; Wehmeyer, R. M.; Rieke, R. D. *J. Org. Chem.*, **1987**, *52*, 5057-5059.
- Gaertner, V. R. *J. Org. Chem.*, **1970**, *35*, 3952-3959.

(Received in Japan 10 June 1997; accepted 3 July 1997)